

## LETTERS TO THE EDITOR

# Hydrophosphoryl Derivatives of Calix[4]resorcinolarenes

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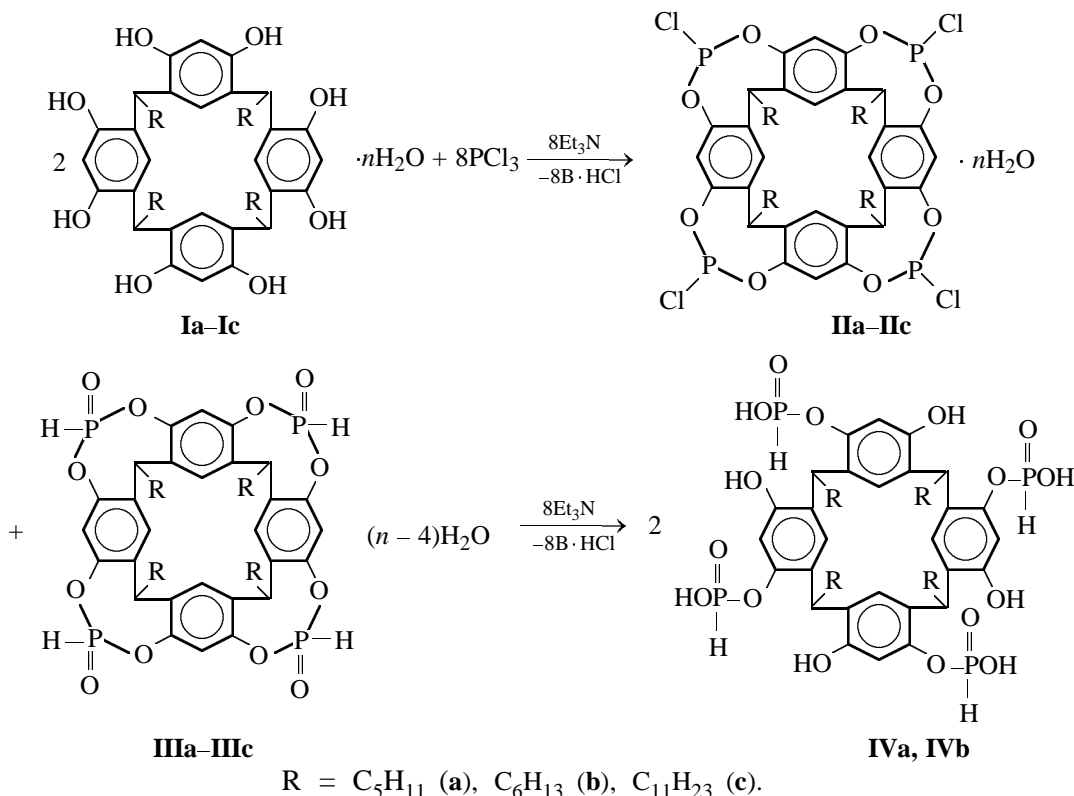
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Recently we showed that the reaction of calyx[4]resorcinolarenes **Ia–c** with phosphorus trichloride in 1:4 molar ratio in the absence of bases leads to formation only of corresponding cyclic phosphorochloridites **IIa–IIc** [1]. It occurred that carrying out this reaction in the presence of 4 moles of triethylamine gives another unexpected synthetic result. Heating of benzene solution of calyxarene **Ia** or **Ib** with phosphorus trichloride and triethylamine in 1:4:4 molar ratio (80°C, 3 h) according to <sup>31</sup>P NMR spectra leads to formation of a mixture of cyclic acid

chlorides and cyclic hydrophosphoryl derivatives of calyx[4]resorcinolarene (**IIa** δ<sub>P</sub> 125.68 ppm; **IIIa** δ<sub>P</sub> –2.35 ppm, <sup>1</sup>J<sub>PH</sub> 758 Hz), (**IIb** δ<sub>P</sub> 125.83 ppm; **IIIb** δ<sub>P</sub> –2.89 ppm, <sup>1</sup>J<sub>PH</sub> 760 Hz) in approximately 1:1 molar ratio. Addition of extra four equivalents of the base to the reaction mixture is accompanied by disappearance of the signals of phosphacyclanes **IIa**, **IIb** and **IIIa**, **IIIb** and formation of linear hydrophosphoryl compounds **IVa** (δ<sub>P</sub> 5.91, <sup>1</sup>J<sub>PH</sub> 714 Hz) or **IVb** (δ<sub>P</sub> 6.53, <sup>1</sup>J<sub>PH</sub> 721 Hz), respectively isolated in a high yield.



Evidently, on the first stage of the process calyxarene **IIa–IIc** react with phosphorus trichloride to form cyclic phosphorochloridite which under the action of intercavit water molecules [2] is converted to cyclic acid **IIIa–IIIc**. Subsequent hydrolysis is accompanied by the cleavage of dioxaphosphocyne fragments with formation of final hydrophosphoryl compounds **IVa** or **IVb**. Note, that hydrolysis of compounds **IIa**, **IIb** and **IIIa**, **IIIb** to the final products **IVa**, **IVb** takes place only in the presence of the latter bases, and the increase in the length of the alkyl radical (compound **III**) improves stability of cyclic hydrophosphoryl compound, and it may be isolated pure. Heating of the mixture of starting reagents in benzene in 1:4:8 molar ratio also leads to formation of incomplete phosphite **IVa** or **IVb**, but in this case it is a problem to isolate them pure because of the presence of small amounts of compounds **II**, **III**. Note, that no data about the preparation of hydrophosphoryl derivatives of calyx[4]resorcinolarenes was published.

**4,6,16,18-Tetrahydroxy-10,12,22,24-tetrakis-(hydroxyhydrophosphoryloxy)-2,8,14,20-tetrapentylcalyx[4]resorcinolarene (IVa)**. A mixture of 0.5 g of calyxarene **Ia**, 0.45 g of phosphorus trichloride, 0.26 g of triethylamine, and 30 ml of benzene were refluxed under argon. Then 0.26 g of triethylamine was added, and the reaction mixture was heated for extra 4 h. Solvent was evaporated, obtained precipitate was filtered off and purified by precipitation from benzene with hexane. Compound **IVa**, 0.6 g (90%) was obtained, mp 82°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 1620 ( $\text{C}_6\text{H}_4$ ), 2430 (P–H), 3250 (OH).  $^1\text{H}$  NMR spectrum ( $\text{CDCl}_3$ ),  $\delta$ , ppm: 0.88 br.m [12 H,  $\text{CH}_3\text{--CH}_2$ ,  $^3J_{\text{HH}}$  6.9 Hz], 1.22 m [24H,  $(\text{CH}_2)_3$ ], 4.40 br.m (4H,  $\text{CH--CH}_2$ ), 6.20 br.s. (4H,  $o\text{-CH}_{\text{arom}}$ ), 9.63 br.s (4H,

OH).  $^{31}\text{P}$  NMR spectrum,  $\delta_{\text{p}}$ , ppm: 5.91 ( $^1J_{\text{PH}}$  714 Hz). Found, %: P 12.10.

**4,6,16,18-Tetrahydroxy-10,12,22,24-tetrakis-(hydroxyhydrophosphoryloxy)-2,8,14,20-tetrahexylcalyx[4]resorcinolarene (IVb)**. It was obtained analogously from 0.5 g of calyxarene **Ib**, 0.42 g of phosphorus trichloride and 0.5 g of triethylamine. Yield 0.54 g (82%), mp 153°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 1620 ( $\text{C}_6\text{H}_4$ ), 2440 (P–H), 3230 (OH).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 0.89 br.m (12H,  $\text{CH}_3\text{--CH}_2$ ,  $^3J_{\text{HH}}$  6.9 Hz), 1.29 br.m [32H,  $(\text{CH}_2)_4$ ], 2.36 m (8H,  $\text{CH}_2\text{--CH}$ ), 4.24 br.m (4H,  $\text{CH--CH}_2$ ), 6.34 br.s (4H,  $o\text{-CH}_{\text{arom}}$ ), 7.39 s (4H,  $m\text{-CH}_{\text{arom}}$ ), 9.69 br.s (4H, OH).  $^{31}\text{P}$  NMR spectrum,  $\delta_{\text{p}}$ , ppm: 6.53,  $^1J_{\text{PH}}$  721 Hz. Found, %: C 56.95; H 7.53; P 10.99.  $\text{C}_{52}\text{H}_{76}\text{O}_{16}\text{P}_4$ . Calculated, %: C 57.78; H 7.04; P 11.48.

IR spectra were recorded on an UR-20 spectrometer in vaseline oil.  $^1\text{H}$  and  $^{31}\text{P}$  NMR spectra were measured on a Bruker WM-250 (250.13 MHz) and Bruker MSL-400 (166.93 MHz) spectrometers against the signals of residual protons of deuterated solvents ( $\text{CDCl}_3$ ) and the external 85% phosphoric acid.

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#### REFERENCES

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